

# Heat Transfer

A layer of ice 0.15m thick has formed on the surface of a deep pond. If the temperature of the upper surface of ice is constant and equal to that of the air which  $-12^{\circ}\text{C}$ , determine the time it will take for the thickness of ice to increase by 0.2mm. Given  $L = 80\text{kcal /kg}$ ,  $\rho = 910 \text{ kg/m}^3$  and  $K = 0.5 \text{ cal /s /m /K}$ .

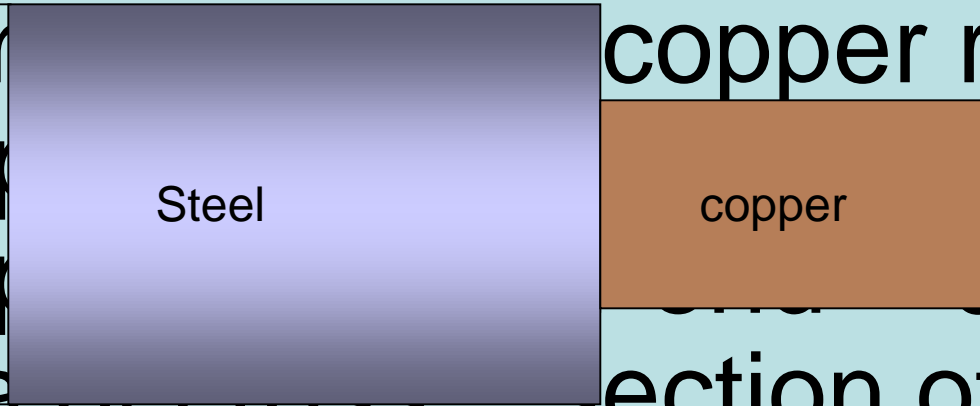
Hint: 
$$dt = \frac{\rho L x dx}{KT} = 364 \text{ s.}$$

Calculate the rate of loss of heat through a glass window of area  $1000\text{cm}^2$  and thickness  $4\text{mm}$  when temperature inside is  $37^\circ\text{C}$  and outside is  $-5^\circ\text{C}$ .

$$K = 0.0022\text{cal/s/cm/K}$$

Ans :  $231\text{ cal/s}$

What is the temp of the steel – copper junction in the steady state of the system shown. Length of steel rod = 15cm, copper rod = 10cm, temp of steel rod = 300°C, temp of copper rod = 0°C. The area of cross section of the steel rod is twice that of the copper rod.  $K_{\text{steel}} = 50.2$  and  $K_{\text{cu}} = 385$  SI units.



Hint : 
$$\frac{K_1 A_1 (300 - T)}{L_1} = \frac{K_2 A_2 (T - 0)}{L_2}$$

# Radiation

# Radiation

The energy emitted by a body in the form of heat by virtue of its temperature is called thermal radiation .

All bodies above absolute zero of temperature emit radiant energy .

Sun is a powerful source of radiant energy .

# Properties of thermal radiation

i. EM

ii.  $3 \times 10^8 \text{ m/s}$

iii. IR  $8000 \text{ \AA} - 20,000 \text{ \AA}$

iv. Reflection, refraction ,  
interference, diffraction,  
polarisation.

Absorptance, Reflectance and Transmittance.

Let  $Q$  – incident radiation

$Q_1$  – qty absorbed,  $Q_2$ - reflected,  $Q_3$   
– transmitted ( $Q = Q_1 + Q_2 + Q_3$ )

Absorbing power or absorptance  $a = Q/Q_1$ .

Reflecting power or Reflectance  $r = Q/Q_2$ .

Transmitting power or transmittance  $t = Q/Q_3$ .

$$a + r + t = 1, \quad \text{if } t = 0, \quad a + r = 1$$

ie. poor reflectors are good absorbers and vice versa.



# Absorptive powers

i) Dull copper – 13%

ii) Indian ink – 85%

iii) Shellac – 72%

iv) Lamp black – 98%

Note:  $a$ ,  $r$  and  $t$  are independent of the nature of material of the body, but depend upon the wavelength of the incident radiation and the nature of the surface.

Absorptance will be different for different wavelengths corresponding to a particular wavelength is called 'monochromatic absorptance' or 'spectral absorptive power'  $a_{\lambda}$ .

# Spectral absorptive power

For a particular  $\lambda$  is defined as the ratio of the heat energy absorbed in a certain time to the total heat energy incident upon it in the same time within a unit wavelength range around the wavelength  $\lambda$ . ie. b/w  $(\lambda - 1/2)$  and  $(\lambda + 1/2)$

Fraction of energy absorbed in the **wave length range  $d\lambda$**   $(\lambda - d \lambda / 2)$  and  $(\lambda + d \lambda / 2)$  =  $a_\lambda d\lambda$  <sup>b/w</sup>

Emissive power 'e'

Of a body at a particular temperature is the total energy radiated per unit sec per unit area.

Spectral emissive power or  
monochromatic emittance  $e_\lambda$

Of a body for a particular  $\lambda$  is defined  
as the quantity of radiant energy  
emitted per sec per unit area of the  
surface within **unit wavelength range**  
around the wavelength  $\lambda$  .

ie. b/w (  $\lambda - 1/2$  ) and (  $\lambda + 1/2$  )

Energy radiated in the **wave length**  
**range  $d\lambda$**  **b/w (  $\lambda - d \lambda / 2$  ) and**  
**(  $\lambda + d \lambda / 2$  )** =  $e_\lambda d\lambda$

Radiant emittance or emissive power 'e'

Radiant emittance  $e$  is the total energy emitted by a body /sec / unit area for all possible wavelengths ranging from 0 to infinity .

$$e = \int e_{\lambda} d\lambda \quad \text{limit 0 to infinity}$$

# Emissivity $\varepsilon$

Is the ratio of the emissive power of a body to the emissive power of a black body at the same temp.

$$\text{Emissivity } \varepsilon = e/E$$

$$\text{Emissive power } e = \varepsilon E$$

# Intensity of the radiation

Is defined as the amount of heat energy falling /sec/ unit area held at right angles to the radiation .

$I \propto 1/r^2$ . inverse square law.

Intensity also depends on the strength of the source.



Spectral intensity  $I_\lambda$

Is defined as the amount of heat energy incident /sec/ unit area held at right angles to the radiation in a unit wave length range around  $\lambda$ .

# Black body

A perfectly black body is not possible in practice. One which absorbs completely the radiations of all wavelengths falling on it hence appears black. (  $\rho = \epsilon = 1$  )  
Very Black body\_ double 1)  
walled hollow sphere having

Black body ( full or total ) radiation

Radiation emitted by a perfectly black body, when it is heated to a suitably high temperature is called black body radiation. (contains all possible wavelengths).

# Prevost's theory of Heat exchange

- i. All bodies above  $0\text{K}$  emit heat to the surroundings and gain heat from the surroundings at all times.
- ii. The amount of radiation emitted increases with temperature.
- iii. Rate of emission depends on the temperature of body and nature of surface.

# Prevost's theory of Heat exchange

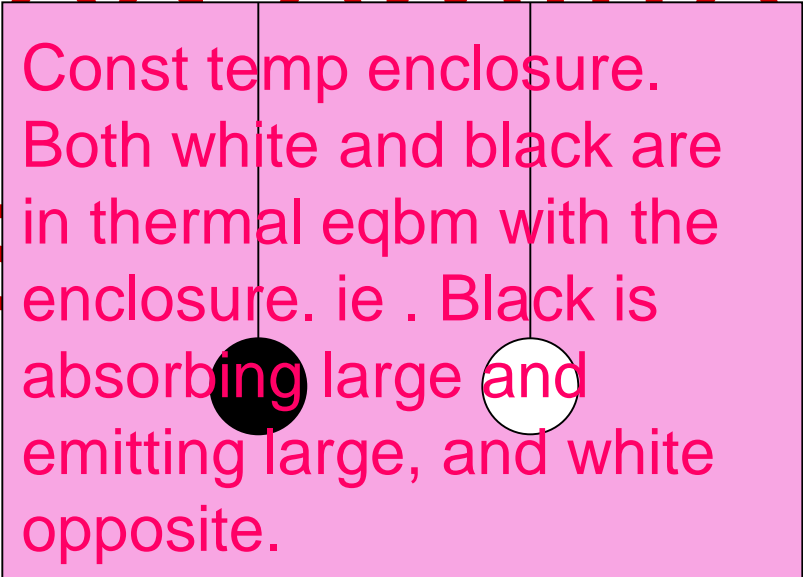
- iv. Rate of emission is not affected by the surroundings.
- v. Continuous exchange of heat b/w body and surroundings.
- vi. Fall or rise in temperature of a body is the net result of exchange of heat between the body and the surroundings.

A body is in thermal equilibrium with its surroundings. Is it static or dynamic equilibrium . Defend.

Dynamic.

Continuous exchange of heat is going on.

Good absorbers are  
Good emitters  
Poor absorbers are  
Poor emitters  
explain.



Const temp enclosure.  
Both white and black are  
in thermal eqbm with the  
enclosure. ie . Black is  
absorbing large and  
emitting large, and white  
opposite.

The diagram shows a pink rectangular enclosure. Inside, a black sphere and a white sphere are suspended by thin vertical lines. The text explains that both spheres are in thermal equilibrium with the enclosure. The black sphere absorbs and emits radiation at a large rate, while the white sphere does the opposite.

Prevost's theory –  
dynamic eqbm

# Kirchhoff's Law

At any given temperature , the ratio of the emissive power to the absorptive power corresponding to a certain wavelength is constant for all bodies and this constant is equal to the emissive power of a perfectly black body at the same temperature and corresponding to the same wavelength.



# Kirchhoff's Law - Proof

Consider a uniformly heated enclosure maintained at temp  $T$ .

$dQ$  – amount of heat having wavelength b/w  $\lambda$  and  $\lambda+d\lambda$  incident on a **unit surface area** /s.

Energy absorbed =  $a_\lambda dQ$

Energy reflected or transmitted

$$= dQ - a_\lambda dQ = dQ (1 - a_\lambda)$$

Energy emitted /s in the wavelength range  $\lambda$  and  $\lambda + d\lambda$  =  $e_\lambda d\lambda$

Total energy coming from the body /sec/unit area(  $\lambda$  and  $\lambda + d\lambda$  ) =  $dQ ( 1 - a_\lambda ) + e_\lambda d\lambda$

Body and enclosure are in thermal eqbm.  
ie. heat coming from body = heat incident.

$$dQ ( 1 - a_\lambda ) + e_\lambda d\lambda = dQ$$

$$e_\lambda d\lambda = a_\lambda dQ \quad (\text{general eqn})$$

For black body  $E_\lambda d\lambda = dQ$  ( $a_\lambda = 1$ )

Ratio gives  $e_\lambda / E_\lambda = a_\lambda$

$$e_\lambda / a_\lambda = E_\lambda \quad \text{Kirchhoff's law.}$$